# Ethyl 3-(2'-deoxyuridin-5-yl)-3-hydroxy-2-iodopropanoate, a nucleoside analogue 

G. Mazumdar, ${ }^{\text {a }}$ M. De, ${ }^{\text {a* }}$ A. Mukhopadhyay, ${ }^{\text {b }}$ S. K. Mazumdar, ${ }^{\text {a }}$ N. Mazumder, ${ }^{\text {c }}$ A. K. Das ${ }^{\text {d }}$ and Edward E. Knaus ${ }^{\text {e }}$

${ }^{\text {a }}$ Department of Materials Science, Indian Association for the Cultivation of Science, Jadavpur, Calcutta 700032, India, ${ }^{\mathbf{b}}$ Department of Physics, University of Jadavpur, Jadavpur, Calcutta 700032, India, ${ }^{\mathbf{c}}$ Department of Physics, Assam University, Silchar, Assam 788015, India, ${ }^{\mathbf{d}}$ Biotechnology Centre, Indian Institute of Technology, Kharagpur 721302, India, and ${ }^{\mathbf{e}}$ Faculty of Pharmacy and Pharmaceutical Sciences, University of Alberta, Edmonton, Alberta, Canada TG6 2N8
Correspondence e-mail: msmd@mahendra.iacs.res.in

Received 10 November 1999
Accepted 18 January 2000
In the title compound, $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{IN}_{2} \mathrm{O}_{8}$, an almost planar heterocyclic base is oriented anti with respect to the puckered sugar moiety. The sugar pucker is $\mathrm{C}^{\prime}$-endo/ $\mathrm{Cl}^{\prime}$-exo, the N glycosidic torsion angle is 166.4 (4) ${ }^{\circ}$ and the conformation of $\mathrm{O}^{\prime}$ ' is $+s c$. The molecules are linked by hydrogen bonds of the types $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$.

## Comment

The title compound, (I), is an example of the 5 -vinyl-2'deoxyuridine derivatives which have been found to inhibit replication of the Herpes simplex virus type I (De Clercq et al., 1979). Studies of the structure-activity relationships of these 5 -substituted derivatives indicated that the activity is associated with analogues in which the 5 -substituent is conjugated with the pyrimidine ring (Griengl et al., 1985). The structural analysis of (I) is a continuation of our earlier work involving systematic conformational studies on modified nucleosides with the aim of understanding structure-function relationships.


(I)

The absolute configuration of the structure has been determined, as indicated by the Flack parameter of -0.04 (3) (Flack, 1983) (Friedel number $=181$ ). The 5 substituent is an aliphatic chain with a terminal carboxyl ethoxy group and it is twisted away from the N1-C6 ring
$\left[\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 7-\mathrm{C} 8\right.$ torsion angle $\left.=-61.6(7)^{\circ}\right]$, allowing the bulky I atom to lie out of the plane of the ring. The chain skeleton has an extended conformation. The terminal ethoxy group has a gauche conformation with respect to the skeleton [C9-O10-C10-C11 = 76.7 ( $\left.{ }^{\circ}\right)^{\circ}$ ]. The $\mathrm{C}-\mathrm{I}$ bond distance of 2.137 (6) A compares closely with the value of $2.139 \AA$ found for $\mathrm{CH}_{3} \mathrm{I}$ (Bowen et al., 1958).
The heterocyclic base moiety is almost planar, the r.m.s. deviation of the atoms from their least-squares mean plane being 0.027 (5) $\AA$ and the maximum deviation being 0.044 (5) $\AA$ for C 2 . The bond lengths and angles for the nucleoside analogue are normal (Allen et al., 1987) with no significant deviation, indicating that the substitution at C5 has no effect on the molecular geometry. The $\mathrm{C} 4=\mathrm{O} 4$ [1.245 (6) $\AA$ ] bond is $0.042(6) \AA$ longer than the $\mathrm{C} 2=\mathrm{O} 2$ $[1.203$ (7) $\AA$ ] and $\mathrm{C} 9=\mathrm{O} 9[1.204$ (7) $\AA$ ] bonds; this elongation is attributed to the formation of a hydrogen bond between O4 and $\mathrm{O}^{\prime} ; \mathrm{O} 2$ and O 9 do not form any hydrogen bonds in the structure of (I). Similar effects have been reported for the 1:1 complex of thiourea and parabanic acid (TUPA; Weber \& Craven, 1987).
The glycosyl torsion angle $\chi\left(\mathrm{C} 2-\mathrm{N} 1-\mathrm{C}^{\prime}-\mathrm{O}^{\prime}\right)$, showing the relative orientation of the base with respect to the sugar ring, has a value of $-193.6(4)^{\circ}$, which lies at the extreme end of the range -160 to $-175^{\circ}$ for torsion angles $\chi$ observed in active anti-HIV nucleosides (Van Roey et al., 1989). This glycosyl angle $\chi$ places the $\mathrm{C1}^{\prime}-\mathrm{O}^{\prime}$ bond close to the plane of the conjugated system of the base, bringing $\mathrm{O}^{\prime}$ ' into close contact with hydrogen H 6 on C 6 of the base [ $\mathrm{H} 6 \cdots \mathrm{O} 4^{\prime}=$ 2.352 (7) Å], resulting in steric hindrance between the base and sugar ring. The structural effects include lengthening of the $\mathrm{Cl}^{\prime}-\mathrm{N} 1$ bond, which has a value 1.482 (7) $\AA$ in (I). Saenger (1984) has correlated the $\mathrm{Cl}^{\prime}-\mathrm{N} 1$ bond length with the angle $\chi$, and for $\chi$ near $-167^{\circ}$, the expected $\mathrm{N} 1-\mathrm{C1}^{\prime}$ bond length is $1.48 \AA$.
The $\mathrm{C1}^{\prime}-\mathrm{O} 4^{\prime}$ bond $\left[1.411(6) \AA\right.$ ] is shorter than $\mathrm{C}^{\prime}-\mathrm{O} 4^{\prime}$ [1.440 (7) $\AA]$, due to the anomeric effect (Kirby, 1983), which is a common feature of nucleosides. The pseudo rotation phase angle $P\left[168.3(6)^{\circ}\right]$ and the maximum torsion angle $\nu_{\text {max }}$ [38.2 (7) ${ }^{\circ}$ ] are calculated from intraring torsion angles and describe the puckering of the five-membered ring of the sugar moiety; these values indicate an unsymmetrical twist of the sugar ring with $\mathrm{C}^{\prime}$-endo/ $\mathrm{C}^{\prime}$ '-exo $\left({ }^{2} T_{3}\right.$; Saenger, 1984) and are similar to those of active anti-HIV compounds, which have $P$ values in the range $165-220^{\circ}$ with $\mathrm{C2}^{\prime}$-endo/ $\mathrm{C}^{\prime}$-exo (Van Roey et al., 1989). The active conformer of $3^{\prime}$-azido- $3^{\prime}$-deoxythymidine (known as AZT) has $P=175^{\circ}$ with $\mathrm{C}^{\prime}$-endo/ $\mathrm{C}^{\prime}$ 'exo (Dyer et al., 1988). The C3'-exo conformation places C5' in an axial position. The torsion angle $\gamma\left[\mathrm{C}^{\prime}-\mathrm{C}^{\prime}-\mathrm{C}^{\prime}-\mathrm{O}^{\prime}=\right.$ $\left.74.9(7)^{\circ}\right]$ describing the orientation of the $5^{\prime}$-hydroxyl group relative to the sugar ring shows that $\mathrm{C5}^{\prime}-\mathrm{O}^{\prime}$ is in a gauchegauche orientation ( $+s c$ ) (Van Roey et al., 1989).

Each molecule of (I) has four H atoms available for hydrogen-bond formation. These are bonded to three hydroxyl O atoms, $\mathrm{O}^{\prime}, \mathrm{O}^{\prime}$ and O 7 , and to the pyrimidine N atom N3. The hydroxyl groups act as both hydrogen-bond donors and acceptors. The hydroxyl groups and the ring-N
atom N 3 form $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds with molecules related by translational symmetry along the $a$ and $b$ axes (Table 2), resulting in a two-dimensional network in the $a b$ plane.


## Figure 1

An ORTEPII (Johnson, 1976) plot of the molecule of (I) drawn at the $30 \%$ probability level. H atoms are drawn as small spheres of arbitrary radii.

## Experimental

Compound (I) was synthesized by the reaction of ethyl 3-(2'-deoxy-uridine-5-yl)propanoate with iodine and potassium iodate, as illustrated in the Scheme, according to the method of Kumar et al. (1989). Crystals of (I) were obtained from methanol by slow evaporation at room temperature.

## Crystal data

$\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{IN}_{2} \mathrm{O}_{8}$
$M_{r}=470.21$
Monoclinic, $P 2_{1}$
$a=6.748$ (3) A
$b=10.466$ (4) $\AA$
$c=12.555$ (4) $\AA$
$\beta=92.36$ (3) ${ }^{\circ}$
$V=885.9(6) \AA^{3}$
$Z=2$

## Data collection

Enraf-Nonius CAD-4 diffractometer
$\omega-2 \theta$ scans
Absorption correction: empirical
via $\psi$ scan (North et al., 1968)
$T_{\text {min }}=0.607, T_{\text {max }}=0.690$
3290 measured reflections
1649 independent reflections
1545 reflections with $I>2 \sigma(I)$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.036$
$w R\left(F^{2}\right)=0.137$
$S=0.928$
1585 reflections
231 parameters
H -atoms: see below
$D_{x}=1.763 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 25
$\quad$ reflections
$\theta=10.3-14.4^{\circ}$
$\mu=1.852 \mathrm{~mm}^{-1}$
$T=293(2) \mathrm{K}$
Block, colourless
$0.35 \times 0.25 \times 0.20 \mathrm{~mm}$

$$
\begin{aligned}
& R_{\text {int }}=0.046 \\
& \theta_{\max }=25^{\circ} \\
& h=-8 \rightarrow 8 \\
& k=0 \rightarrow 12 \\
& l=0 \rightarrow 14 \\
& 3 \text { standard reflections } \\
& \quad \text { every } 100 \text { reflections } \\
& \text { intensity decay: negligible }
\end{aligned}
$$

Table 1
Selected torsion angles ( ${ }^{\circ}$ ).

| $\mathrm{C} 5-\mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9$ | $-172.8(5)$ | $\mathrm{C} 8-\mathrm{C} 9-\mathrm{O} 10-\mathrm{C} 10$ | $-175.3(7)$ |
| :--- | :--- | :--- | ---: |
| $\mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9-\mathrm{O} 10$ | $-139.7(6)$ | $\mathrm{C} 9-\mathrm{O} 10-\mathrm{C} 10-\mathrm{C} 11$ | $76.7(9)$ |

Table 2
Hydrogen-bonding geometry ( $\AA \AA^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O} 5^{\prime}-\mathrm{H}^{\prime} \cdots \cdots 3^{\prime \prime}$ | 0.82 | 2.01 | 2.732 (6) | 147 |
| $\mathrm{O} 7-\mathrm{H} 7 \cdots \mathrm{O}^{\prime \text { 'ii }}$ | 0.82 | 2.29 | 2.912 (7) | 133 |
| $\mathrm{O} 3^{\prime}-\mathrm{H3}^{\prime} \cdots \mathrm{O} 4^{\text {iii }}$ | 0.82 | 1.98 | 2.650 (7) | 138 |
| $\mathrm{N} 3-\mathrm{H} 3 \cdots \mathrm{O} 7^{\text {iv }}$ | 0.86 | 2.07 | 2.860 (6) | 152 |

Symmetry codes: (i) $x-1, y, z$; (ii) $x, y-1, z$; (iii) $x, 1+y, z$; (iv) $1+x, y, z$.

The number of Friedel-related reflections was 181. All H atoms attached to C or N were located geometrically and refined as part of a riding model, with $U_{\text {iso }}(\mathrm{H})=0.078(6) \AA^{2}$. All H atoms attached to hydroxyl-O atoms were located in circular difference Fourier syntheses and thereafter refined freely with a common $U_{\text {iso }}(\mathrm{H})$ of 0.078 (6) $\AA^{2}$. The atom-numbering scheme used in Fig. 1 is consistent with the rules of the IUPAC-IUB Joint Commission on Biochemical Nomenclature (IUPAC-IUB, 1983).

Data collection: CAD-4 Software (Enraf-Nonius, 1989); cell refinement: CAD-4 Software; data reduction: CAD-4 Software; program(s) used to solve structure: SHELXS86 (Sheldrick, 1990); program(s) used to refine structure: SHELXL93 (Sheldrick, 1993); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL93.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: BM1390). Services for accessing these data are described at the back of the journal.

## References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. \& Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-S19.

Bowen, H. P. M., Donohue, J., Jenkin, D. G., Kennard, O., Wheatley, P. J. \& Whiffen, D. H. (1958). Table of Interatomic Distances and Conformations in Molecules and Ions, edited by L. E. Sutton, p. M112. London: The Chemical Society.
De Clercq, E., Descamps, J., De Somes, P., Barr, P. J., Jones, A. S. \& Walker,
R. T. (1979). Proc. Natl Acad. Sci. USA, 76, 2947-2951.

Dyer, I., Low, J. N., Tollin, P., Wilson, H. R. \& Howie, R. A. (1988). Acta Cryst. C44, 767-769.
Enraf-Nonius (1989). CAD-4 Software. Version 5.0. Enraf-Nonius, Delft, The Netherlands.
Flack, H. D. (1983). Acta Cryst. A39, 876-881.
Griengl, H., Bodenteich, M., Hayden, W., Wanck, E., Streicher, W., Stutz, P., Bachmayer, H., Ghazzouli, I. \& Rosenwirth, B. (1985). J. Med. Chem. 28, 1679-1684.
IUPAC-IUB Joint Commision on Biochemical Nomenclature (1983). Eur. J. Biochem. 131, 9-15.
Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
Kirby, A. J. (1983). In Anomeric Effects and Related Stereoelectronic Effects at Oxygen, pp. 10-21. Berlin: Springer-Verlag.
Kumar, R., Wiebe, L. I., Hall, T. W., Knaus, E. E., Tovell, D. R. \& Tyrrel, D. L. (1989). J. Med. Chem. 32, 941-944.

North, A. C. T., Phillips, D. C. \& Mathews, F. S. (1968). Acta Cryst. A24, 351359.

Saenger, W. (1984). In Principles of Nucleic Acid Structure, pp. 9-104. New York: Springer-Verlag.
Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
Sheldrick, G. M. (1993). SHELXL93. University of Göttingen, Germany.
Van Roey, P., Selerno, J. M., Chu, C. K. \& Schinazi, R. F. (1989). Proc. Natl Acad. Sci. USA, 86, 3929-3933.
Weber, H. P. \& Craven, B. M. (1987). Acta Cryst. B43, 202-209.

